

This article was downloaded by:

On: 23 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713617200>

Synthesis of Functionalized Thiodisaccharides by Conjugate Addition

Bernd Becker^a; Julian Thimm^a; Joachim Thiem^a

^a Institut für Organische Chemie, Martin-Luther-King-Platz, Hamburg, Germany

To cite this Article Becker, Bernd , Thimm, Julian and Thiem, Joachim(1996) 'Synthesis of Functionalized Thiodisaccharides by Conjugate Addition', Journal of Carbohydrate Chemistry, 15: 9, 1179 – 1181

To link to this Article: DOI: 10.1080/07328309608006505

URL: <http://dx.doi.org/10.1080/07328309608006505>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

COMMUNICATION

SYNTHESIS OF FUNCTIONALIZED THIODISACCHARIDES BY CONJUGATE ADDITION

Bernd Becker, Julian Thimm and Joachim Thiem*

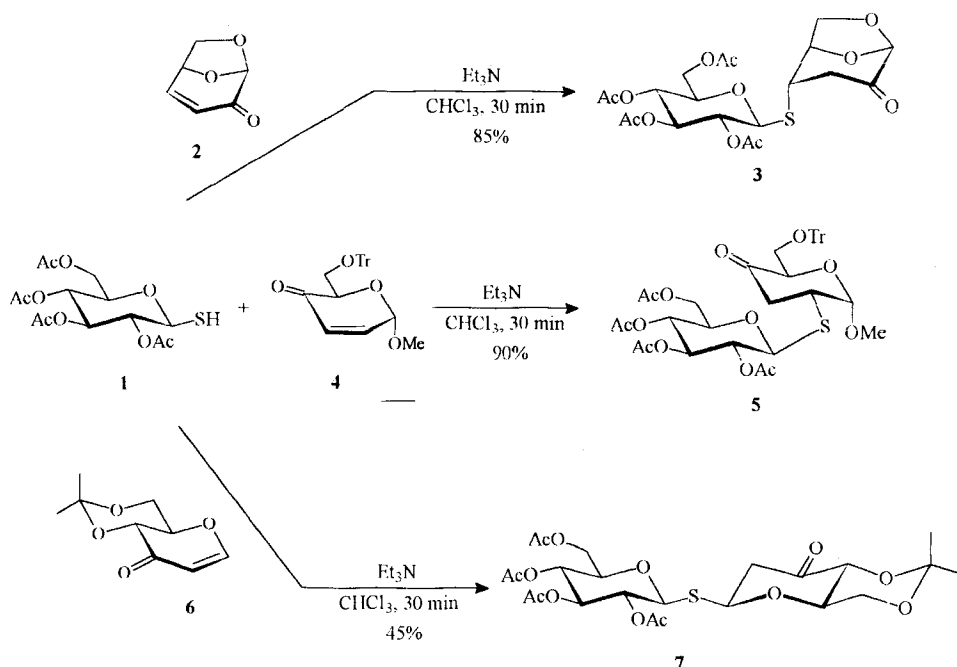
Institut für Organische Chemie, Martin-Luther-King-Platz 6, D-20146 Hamburg, Germany

Received April 23, 1996 - Final Form August 31, 1996

Oligosaccharides containing thioglycosidic linkages are resistant to enzyme-catalysed hydrolysis and therefore interesting material for studies of carbohydrate hydrolases. Some of these were shown to be competitive inhibitors for several glycanases such as α -amylase,¹ cellobiohydrolase I and II,² and for α -L-fucosidase,³ a glycosidase. Thio-oligosaccharides have also been used to prepare column material for affinity chromatography to isolate carbohydrate hydrolases. In contrast, affinity material containing normal *O*-glycosidic bonds was hydrolysed and led to column deterioration. By employing this approach, affinity materials containing 1,4-dithiocollobose and 1,4,4'-trithiocellotriose could be used successfully to separate the cellobiohydrolases of *Trichoderma reesei*.²

Whereas the standard synthesis of thio-oligosaccharides proceeds via a nucleophilic displacement, we now present a simple method for the preparation of functionalized thio-disaccharides by conjugate addition of 2,3,4,6-tetra-*O*-acetyl-1-thio- β -D-glucose (1)⁴ to unsaturated carbohydrate derivatives. Thiols easily react with unsaturated acceptor systems in 1,4-additions as has been demonstrated in various examples.⁵ The addition of alkyl thiols^{6, 7} and thiophenol⁸ to levoglucosenone (2) with triethylamine as catalyst has been reported to give excellent yields and recently the synthesis of a 1-thio- α -L-fucopyranoside has also been reported.⁹

The addition of 2,3,4,6-tetra-*O*-acetyl-1-thioglucose (1) to levoglucosenone¹⁰ proceeds in chloroform solution with only traces of triethylamine as catalytic base¹¹ to yield 85% of the adduct 3¹² after recrystallisation from toluene. Apparently, due to the steric hindrance created by the 1,6-anhydro bridge a nucleophilic attack is possible only from the opposite face of the molecule, thus forming the axial product exclusively.



Similarly, nucleophilic attack of 1 to hex-2-enopyranos-4-yl-*O*-trityl- β -D-glucopyranoside 4^{13, 14} led to a single addition product 5¹² in 90% yield after flash chromatography. In this case the bulky trityl group is assumed to prevent an attack from the upper side of 4 which could give the alternative axial addition product.

In the acceptor system of hex-1-enopyranos-3-yl-*O*-tert-butyl- β -D-glucopyranoside 6,^{14, 15} the addition of compound 1 led to the product 7,¹² forming preferentially the β -thioglycosidic bond. The lower yield observed in the preparation of this trehalose type disaccharide 7 (45% after flash chromatography) is considered to be caused by the instability of the 2-deoxythioglycoside and competing retro reaction.

ACKNOWLEDGEMENT

We wish to thank Dr. Regine Blattner from Industrial Research Limited, Lower Hutt, New Zealand for the generous gift of levoglucosenone.

REFERENCES AND NOTES

1. M. Blanc-Muesser, L. Vigne and H. Driguez, *Carbohydr. Res.* **224**, 59 (1992).
2. C. Orgeret, E. Seillier, C. Gautier, J. Defaye and H. Driguez, *Carbohydr. Res.* **224**, 29 (1992).
3. H. Hashimoto, K. Shimada and S. Horito, *Tetrahedron: Asymmetry* **12**, 2351 (1994).
4. D. Horton, *Methods Carbohydr. Chem.* **2**, 433 (1963).
5. P. Perlmutter, *Conjugate Addition Reactions in Organic Synthesis*, Pergamon Press, Oxford, 1992 and references cited therein.
6. M. G. Essig, *Carbohydr. Res.* **156**, 225 (1986).
7. B. Becker, Dissertation, Universität Hamburg, 1996.
8. F. Shafizadeh, R. H. Furneaux and T. T. Stevenson, *Carbohydr. Res.* **71**, 169 (1979).
9. Z. J. Wiczak, J. Sun and R. Mielguj, *Bioorg. Med. Chem. Lett.* **5**, 2169 (1995)
10. Z. J. Wiczak *VIIth European Carbohydrate Symposium*, Cracow, Poland, 1993, Abstract A103.
11. Typical procedure: The saccharide acceptor (0.5 mmol) and the thioglucose (1, 200 mg, 0.57 mmol) were stirred in chloroform (10 mL) at room temperature and triethylamine (2 μ L) was added. Stirring was continued for 30 min, the solvent evaporated and the raw material purified by crystallisation or chromatography.
12. Representative physical data:
(3) mp 161-164 °C, $[\alpha]_D^{20}$ -125° (c 1, chloroform), $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 5.14 (s, 1 H, H-1), 4.59 (d, 1 H, $J_{1,2} = 10.2$ Hz, H-1'), 3.98-4.06 (m, 2 H, H-6endo and H-6exo), 3.58 (m, 1 H, H-4), 3.09 (dd, 1 H, $J_{3eq,4} = 8.1$ Hz, $J_{gem} = 17.8$ Hz, H-3eq), 2.04 (dd, 1 H, $J_{3ax,4} = 1.5$ Hz, H-3ax).
(5) mp 56 °C, $[\alpha]_D^{20}$ +43° (c 1, chloroform), $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 4.99 (d, 1 H, $J_{1,2} = 4.4$ Hz, H-1), 4.65 (d, 1 H, $J_{1,2} = 10.1$ Hz, H-1'), 4.11 (dd, $J_{5,6a} = 1.9$ Hz, H-5), 3.43 (s, 3 H, OMe), 3.41 (dd, 1-H, $J_{gem} = 9.2$ Hz, H-6a), 3.32 (dd, 1 H, H-6b), 3.25 (dt, 1 H, $J_{2,3ax} = 8.2$ Hz, H-2), 2.61 (m, 2 H, H-3ax and H-3eq);
(7) $^1\text{H NMR}$ (400 MHz, CDCl_3): δ = 5.16 (dd, 1 H, $J_{1,2ax} = 12.2$ Hz, $J_{1,2eq} = 3.1$ Hz, H-1), 4.65 (d, 1 H, $J_{1,2} = 9.6$ Hz, H-1'), 4.38 (d, 1 H, $J_{4,5} = 10.2$ Hz, H-4), 4.06 (dd, 1 H, $J_{5,6a} = 5.6$ Hz, $J_{gem} = 10.2$ Hz, H-6a), 3.94 (dd, 1 H, $J_{5,6b} = 10.2$ Hz, H-6b), 3.59 (ddd, 1 H, H-5), 2.80 (dd, 1 H, $J_{gem} = 14.8$ Hz, H-2ax), 2.71 (dd, 1 H, H-2eq), 1.52 (s, 3 H, CH_3), 1.50 (s, 3 H, CH_3).
13. B. Fraser-Reid, D. L. Walker, S. Y. K. Tam and N. L. Holder, *Can. J. Chem.* **51**, 3950 (1973).
14. S. Kim and R. G. Salomon, *Tetrahedron Lett.* **30**, 6279 (1989).
15. S. Czernecki, K. Vijayakumaran and G. Ville, *J. Org. Chem.* **51**, 5472 (1986).